Course Title/Number	Principles of Chemistry Laboratory I / CHM 217 Sections 107 & 108	
Semester/Year	Fall 2017	
Days/Time	Wednesday 2:00-4:50 PM	
Location	465 Science Hall (pre-lab lecture), 474/476 (laboratory)	
Instructor	Dr. Bin Wang	
Office	241L Byrd Biotechnology Science Center	
Phone	(304) 696-3456	
E-Mail	wangb@marshall.edu	
Office Hours	Tuesday 1:00-4:00 PM (BBSC 241L), Thursday 1:30-3:30 PM (L.A. Session room, S 460), or by appointment	
University Policies	By enrolling in this course, you agree to the University Policies listed below. Please read the full text of each policy by going to  http://www.marshall.edu/academic-affairs/policies/ 	

# **Course Description:**

A laboratory course that demonstrates the application of concepts introduced in CHM 211. 2 credit hours. Corequisite or prerequisite: CHM 211

# **Required Texts, Additional Reading, and Other Materials:**

- 1. CHM 217 Lab Manual
- 2. access to MUOnLine and a Marshall email account
- 3. composition notebook (not spiral-bound) and blue/black ink pen
- 4. non-programmable calculator
- 5. indirectly vented chemical safety goggles
- 6. roll of paper towels for cleanup
- 7. combination lock for lab drawer (optional)

Student Learning Outcomes	How students will practice each outcome in this course	How each outcome will be assessed
Students will know and follow safety rules in the chemical laboratory.	<ul> <li>safety training at MUOnLine</li> <li>reading laboratory manual</li> </ul>	<ul> <li>online safety quiz</li> <li>midterm and final exams</li> <li>instructor's evaluation</li> </ul>
Students will learn how to properly use and care for laboratory equipment.	<ul> <li>reading laboratory manual</li> <li>prelab lecture</li> <li>laboratory experiments</li> </ul>	<ul><li> lab reports</li><li> instructor's evaluation</li></ul>
Students will learn how to record and communicate laboratory experiments and results.	<ul> <li>reading laboratory manual</li> <li>prelab lecture</li> <li>laboratory experiments</li> </ul>	<ul><li> lab notebook</li><li> lab reports</li></ul>

Students will apply concepts introduced in chemistry lecture (CHM 211).	<ul> <li>reading laboratory manual</li> <li>laboratory experiments</li> <li>laboratory calculations</li> </ul>	<ul><li> pre- and post-lab questions</li><li> midterm and final exams</li></ul>
(CIIIVI 211).	• laboratory calculations	

#### **Grading Policies:**

lab reports (including post-lab questions)	55	points
quizzes (8 during the semester)	10	points
midterm exam	13	points
final exam	15	points
lab notebook & pre-lab questions	5	points
instructor evaluation	2	points
	100	TOTAL POINTS

**Grading Scale:** A: 90-100, B: 80-89, C: 70-79, D: 60-69, F: < 60

# **Attendance Policy:**

Attendance is required to complete and receive credit for experiments. The Department of Chemistry policy requires that students complete at least 75% of laboratories. Students will receive a grade of "F" for missing 4 or more laboratories, whether the absences are excused or unexcused. Students with excused absences must contact the instructor as soon as they are permitted to return to campus in order to schedule a make-up lab. Do not wait until the following week's lab to make arrangements. If you anticipate missing a lab, notify the instructor as soon as possible. It may be possible to make *prior arrangements* to complete the lab in the same week with a different section and instructor. *Permission from Dr. Wang and the alternate instructor is required*. If class is cancelled unexpectedly, scheduled assignments will be due and scheduled exam will be given during the next class meeting.

# Lab Safety:

The safety rules for the laboratory are outlined in your lab manual. Safety training must be completed at MUOnLine prior to the given deadline or the student will be denied admission to the lab. Proper clothing is of the utmost importance. This means that legs should be covered down past the knees (pants are best) and the entire torso should be covered. Shoes must completely cover feet, including the tops of the feet. (No ballet flats or sandals.) If your attire is unsafe, you must change before entering the lab. Any reckless disregard for safety (horseplay, frequent/willful lapses in wearing of goggles, etc.) may result in dismissal from the lab and failure of the course. Cell phones and similar electronic devices should not be used while in the lab except as calculators and timing devices. Misuse of electronics may result in dismissal from lab for the day and a failing grade for the experiment. The ACS academic lab safety guide is available to download at <a href="http://www.acs.org/content/dam/acsorg/about/governance/committees/chemicalsafety/publications/safety-in-academic-chemistry-laboratories-students.pdf">http://www.acs.org/content/dam/acsorg/about/governance/committees/chemicalsafety/publications/safety-in-academic-chemistry-laboratories-students.pdf</a>

#### **Miscellaneous Policies:**

Please silence cell phone ringers during class or exams. Use of cell phones / PDAs / MP3 players and similar devices during exams will be considered academic dishonesty. Class announcements may occasionally be made via email to your university email address. Please check it on a regular basis. Lecture slides will be posted at MUOnLine.

	Pre-lab Lectures	Lab Activities		
Week 1	Syllabus	Lab Check-In, Density of Water		
8/23	Exp. 1 Part I	H1 and H2: sig figs and dimensional analysis		
Week 2	Erra 1 De at II	Determination of Sugar in Soft Drinks &		
8/30	Exp. 1 Part II	Graphing with Excel		
Week 3	Exp. 2	Segreting the Components of a Hotors are and Minture		
9/6	Quiz 1	Separating the Components of a Heterogeneous Mixture		
Week 4	Exp. 3	Determination of the Persont Ownson in Air		
9/13	Quiz 2	Determination of the Percent Oxygen in Air		
Week 5	Exp. 4	Determination of an Empirical Formula		
9/20	Quiz 3	Determination of an Empirical Formula		
Week 6	Exp. 5	Determination of Avagadra's Number		
9/27	Quiz 4	Determination of Avogadio's Number		
Week 7	Exp. 6	Synthesis of an Alum		
10/4	Midterm Exam	Synthesis of an Alum		
Week 8	$E_{\rm YP}$ 7	Deastions		
10/11	Exp. 7	Reactions		
Week 9	Exp. 8	Determination of Molar Mass		
10/18	Quiz 5	Determination of Molar Mass		
Week 10	Exp. 9	Heat of Peaction and Heat of Solution		
10/25	Quiz 6	field of Reaction and field of Solution		
10/27 is last day to drop a full semester individual course				
Week 11	Exp. 10	The Titration of Vinegar		
11/1	Quiz 7	The Thration of Vinega		
Week 12	Exp. 11	Combustion! Synthesis and Reactions of Oxygen		
11/8	Quiz 8	Combustion:-Synthesis and Reactions of Oxygen		
Week 13	$E_{\rm res}$ 12	Energy in a Peanut: Calorimetry		
11/15	Exp. 12			
Week 14	Thankaciving Preak			
11/22	Гпапкѕдічіпд Вгеак			
Week 15	Exp. 13	Malagular Architecture Lab Chasteret		
11/29	Review	Molecular Architecture, Lab Checkout		
Week 16	Final Exam			
12/6				