

CHM-211: Principles of Chemistry I
Fall 2014
Course Syllabus
Section 105

Lecture Instructor: Dr. Rosalynn Quiñones
Science Building, Room 496
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Lecture Meeting

Times: Science Building 473 MWF 1:00 – 1:50 pm

Course Credits: 3

Office Hours: MWF 9:00 am–10:00 am or by appointment. I welcome drop-in visits, but cannot guarantee that I will be available to help you during non-office hours. Simple questions can be answered via email.

Prerequisites: Mathematics ACT of 23 or higher (Math SAT of 540) or consent of instructor

Course Overview:

Welcome to Principles of Chemistry! In this course, the students will be introduced to the principles behind chemistry. The purpose of this course is providing the necessary background to choose the most appropriate instrumental method to solve qualitative and quantitative chemical problems efficiently.

One of the main objectives of this course is to promote active learning that will be accomplished by having students using “clickers” questions in the classroom. Problem solving skills and critical thinking will be improved by engaging students solving activities in lecture, and learn to search and read the chemical literature in order to address problems associated with chemistry understanding the world we see, and the world of atoms and molecules.

My main goal is to turn students in to scientist and not train them to be technician. By doing so, students become more enthusiastic about learning and understanding chemical theories.

Student Learning Objectives:

Students will:

1. Understand the chemical theories related to: atomic theory, chemical bonding, kinetic molecular, and intermolecular forces.
2. Apply these theories to problems in order to explain chemical realities.
3. Acquire problem solving skills during different assessments; in groups and individually.
4. Obtain, utilize, and assimilate information from the body of work known as the chemical literature.
5. Mathematical assess qualitative and quantitative data obtained by various tools.

Materials

Lecture: Text: Chemistry: The Science in Context, 3rd Edition by Gilbert, Kirss, Foster, and Davies; a Subscription to Smartwork online homework; Scientific Calculator; Notebook; clicker

Examination: Scientific Calculator; Pencils. No Calculators with alphanumeric and/or graphing capabilities are not permitted for exams or the final exam. If you have questions regarding your calculator, I will be glad to look at it. Make sure you do this before the day of a test. Also, you may not use your cell phone as a calculator.

Attendance

Attendance for this course is optional, but strongly encouraged. Absences from exams can only be made-up if the absence falls within one of the categories outlined in the undergraduate catalog. To make-up an exam, you will need to follow the process for securing an excused absence. All excused absences must be obtained as soon as possible. You are responsible for all announcements and material given during class. A tentative lecture schedule of topics is attached.

Course Policy

Course Assessments

Exams: There are 4 (four) exams which will be given during the lecture period. The exams will consist of multiple choices, essay questions, and problem solving in which all steps leading to the solution must be shown.

Final Exam: The final exam is a cumulative departmental exam. This exam will be given during final exam period. Re-grades will be considered within one day of the exam being returned. If you request a re-grade, the entire assignment will be reviewed. This may result in either the loss or gain of points.

Clickers will be part of lecture assessment and are required in lecture every day. Course points for clickers' questions will be awarded for participation. There are no make-ups for missed clicker questions. You can purchase a clicker from the bookstore or you can purchase one from the online store through Turning Technologies, our university-adopted clicker company. You can buy the clicker from this website (\$32) or buy one in the bookstore. You need to have an actual clicker purchased by the beginning of week 2 or you will not obtain clicker points for the days you do not have a clicker. To access the on-line store:

Visit: store.turningtechnologies.com

The school code for Marshall is: mar1

Homework is very important and useful to enhance the comprehension of the material. Homework will be assigned with each chapter and will be submitted online using SmartWork. Each homework assigned by chapter will count as 5 points towards the cumulative of 50 points. Some homework will be discussed by the student during the lecture to promote active learning. It will be *very difficult* to do well in this class without doing the homework. Doing practice problems is the best way to learn the material.

Late work: Late work will be penalized 25% less per day. No assignments will be accepted more than 4 days late.

STUDENT SELF-REGISTRATION

Students have to create a SmartWork account and self-enroll into your class following the "First Time User" instructions at <http://smartwork.wwnorton.com>

Students will need:

1. A valid email address
2. The enrollment key for your course: **CHEM46608 (case sensitive)**
3. A registration code from W.W. Norton. This proof-of-purchase allows students to access your course after their free two-week trial period expires.

Course Point Allocation	
4- 1 hour exams (100 points each)	400 pts
Online homework (50 points)	50 pts
"Clickers" questions (50 points)	50 pts
Final cumulative exam	200 pts
Total	700 pts

Grade Scale

Grade Chart	
100-90	A
89-80	B
79-70	C
69-60	D
59-0	F

Miscellaneous policies

I have an Open Communication Policy: If you are having trouble with a problem, concept, or anything class related please do not hesitate to email me or come by my office. Please silence cell phone ringers during class or exams. The instructor reserves the right to answer any ringing cell phones during lecture, or to dismiss the offending student. Use of cellphones / PDAs / MP3 players and similar devices during tests and exams will be considered academic dishonesty. Recording of lectures without the instructor's permission is prohibited. Laptops should not be used during class without permission. The content of this course will adhere closely to the information contained in the textbook. You may use other resources (alternate texts, notes from other professors, etc.). If you find information that contradicts something written in the textbook or said in the lecture, please consult Dr. Quiñones. A "Teachers.io" online account was created for this section. Students can download for free "MyHomework student planner" app using a smartphone. Class announcements, resources, and lessons will be posted each week in this app.



Class announcements may occasionally be made via email to your university email address. Please check it on a regular basis. Lecture notes and handouts will be posted at MU Online as time permits.

Academic Honesty

The university policy will be enforced. See page 70 of the 2014-15 undergraduate catalog. Some examples of academic dishonesty include (but are not limited to) copying another student's assignment, lying about being ill on the day of a test, using a cell phone or other communication device during a test, quoting an author's writing (including material found on the internet) without giving due credit.

http://www.marshall.edu/ucomm/files/web/UG_14-15_published.pdf

Incomplete Coursework

The university policy will be enforced. See page 93 of the 2014-15 undergraduate catalog.

http://www.marshall.edu/ucomm/files/web/UG_14-15_published.pdf

D/F Repeat Rule

See page 88 of the 2014-15 undergraduate catalog.

http://www.marshall.edu/ucomm/files/web/UG_14-15_published.pdf

Accommodations for Disabilities

Students with disabilities must contact the Office of Disabled Student Services in Prichard Hall 117, phone 696-2271 to provide documentation of their disability to ensure proper accommodation. Please visit <http://www.marshall.edu/disabled> for additional information.

Tentative Course Schedule

Weeks	Lecture
Aug. 25- 29	Chapter 1
<i>Sept. 1</i>	<i>Labor day-No classes</i>
Sept. 2 - 5	Chapters 2
Sept. 8 -12	Chapters 3
<i>Sept 12</i>	<i>Exam 1 (Chapters 1-3)</i>
Sept. 15-19	Chapter 4
Sept. 22-26	Chapter 4, 5
Sept. 29- Oct. 3	Chapters 5
<i>Oct. 3</i>	<i>Exam 2 (Chapters 4-5)</i>
Oct. 6-10	Chapter 6
Oct. 13-17	Chapter 6
Oct. 20-24	Chapter 7
Oct. 27- Oct. 31	Chapter 8
<i>Oct. 31</i>	<i>Exam 3 (Chapters 6-7)</i>
Nov. 3- 8	Chapter 8, 9
Nov. 10- 14	Chapter 9
Nov. 17-21	Chapter 10
<i>Nov. 24- 28</i>	<i>Thanksgiving break – No classes</i>
Dec. 1 - 5	Chapter 10, 11
<i>Dec. 3</i>	<i>Exam 4 (Chapters 8-10)</i>
<i>Dec. 5</i>	<i>Last day of Classes</i>
Dec. 6 (Saturday)	Final Exam- 10:00 am

**** This schedule is subject to change. Changes, if necessary, will be announced in class****