CHEM 345: Introduction to Analytical Chemistry

Fall 2013

Course Instructor:

Dr. Bin Wang

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Course Prerequisites: CHM 212 and 218

Lectures: S-473, Tuesday & Thursday 12:30-1:20 pm

Textbook: Quantitative Chemical Analysis, 8th edition, by Daniel C. Harris

ACS Academic Lab Safety Guide:

 $\underline{http://portal.acs.org/portal/PublicWebSite/about/governance/committees/chemicalsafety/publicated about/governance/committees/chemicalsafety/publicated about/governance/chemicalsafety/publicated about/governance/chemicalsafety/governance/chemic$

ions/WPCP 012294

Labs: S-492, Tuesday & Thursday 1:30-3:15 pm

Lab Requirements: Lab goggles and a bound laboratory notebook

Course Description:

Introduction to the basic principles of Analytical Chemistry including traditional wet methods and contemporary instrumental methods of chemical analysis.

Course Objectives:

- 1. To learn how to analyze results through statistical methods.
- 2. To learn chemical equilibrium, titrations, and basic knowledge of electrochemistry.
- 3. To develop wet laboratory techniques essential for high precision experimentation.
- 4. To acquire the ability to operate advanced instrumentation and to interpret results through modern theory.

Grading: Homework 20%, Quizzes 15%, Labs 40%, and Final Exam 25%

Grading Scale: A 90-100 B 80-89 C 70-79 D 60-69 F < 60

Attendance:

Attendance for this class is not mandatory. Absences from quizzes and laboratories can only be made-up if the absence falls within one of the categories outlined in the undergraduate catalog. To make-up a quiz or lab, you will need to follow the process for securing an excused absence. All excused absences must be obtained as soon as possible.

Course Policies:

- 1. Homework and lab reports will not be accepted after their due dates.
- 2. Graphing calculators, calculators with alphanumeric programming, and calculators on cell phones, PDAs, etc. cannot be used during quizzes/exams. Likewise, sharing of calculators during quizzes/exams is prohibited.
- 3. During quizzes/exams, all materials necessary will be provided to you except a pencil and calculator. You may NOT use your own paper, etc.
- 4. Please turn off cell phones during class, failure to do so may result in dismissal from lecture.

- 5. Students with disabilities who require special accommodations will be made. www.marshall.edu/disabled.
- 6. Academic dishonesty will be dealt with as outlined in the undergraduate catalog.

Tentative Schedule:

	Tuesday		Thursday	
Week 1	Chapter 0,1,2	HW	Chapter 3,4	HW
8-27, 8-29	Lab: Balance (check in)		Lab: Volumetric Tech	
Week 2	Chapter 4	HW	Chapter 26	HW
9-3, 9-5	Lab: Grav. Cl		Lab: Grav. Cl	
Week 3	Chapter 26	Quiz	Chapter 5,6	HW
9-10, 9-12	Lab: Grav. Cl		Lab: Stand. NaOH	
Week 4	Chapter 6,7	HW	Chapter 7	Quiz
9-17, 9-19	Lab: % KHP		Lab: Stand. HCl, Soda Ash	
Week 5	Chapter 8	HW	Chapter 9	HW
9-24, 9-26	Lab: Soda Ash Unknown		Lab: pH Meter	
Week 6	Chapter 10	HW	Chapter 11	Quiz
10-1, 10-3	Lab: Titration Curve		Lab: Unknown Acid	
Week 7	Chapter 13,15	HW	Chapter 13,14	HW
10-8, 10-10	Lab: Redox %Fe Known		Lab: Redox %Fe Unknown	
Week 8	Chapter 14	Quiz	Chapter 17,18 Intro. Spectroscopy	HW
10-15, 10-17	Lab: [Cl ⁻] and the Nernst Equ.		No Lab	
Week 9	Chapter 17,18,19 Beer's Law	HW	Chapter 17,19 Vibrational Spec.	HW
10-22, 10-24	Lab: UV/Vis		Lab: UV/Vis	
Week 10	Chapter 17 Luminescence	Quiz	Chapter 20	HW
10-29, 10-31	Lab: IR		Lab: IR	
Week 11	NMR Spectroscopy	HW	Chapter 21	Quiz
11-5, 11-7	Lab: Fluorescence		Lab: Fluorescence	
Week 12	Chapter 22	HW	Chapter 23	HW
11-12, 11-14	Lab: Paper Chromatography (chec	k out)	Lab: GC	
Week 13	Chapter 24	Quiz	Chapter 24	HW
11-19, 11-21	Lab: GC		Lab: HPLC	
Week 14	Thanksgiving/Fall Break		Thanksaiving/Fall Proak	
11-26, 11-28	Thanksgiving/Fatt Break		Thanksgiving/Fall Break	
Week 15	Chapter 25	HW	Review	Quiz
12-3, 12-5	Lab: HPLC			
Week 16	Final Exam, 12:45 - 2:45			
12-10				

Suggested Activities for Success:

- 1. Read the suggested material from the textbook before and after each lecture.
- 2. Try to work through every homework problem no matter how difficult.
- 3. Always attend class and take good notes.
- 4. Seek help from others, some possibilities:
 - a. Take advantage of office hours.
 - b. Work in small groups on studying for quizzes and the final.
- 5. Exercise regularly and maintain a healthy diet.