

Syllabus

Chemistry 217, Sections 201 and 202 Spring 2014

Instructor: Dr. Lawrence Schmitz **Phone:** 304 696-2373 **Office:** S-480 **e-mail:** schmitz@marshall.edu

Office Hours: MTWRF 1-2 or by appointment

Credit: 2 hours

Pre- or Co- requisite: CHM 211

Class Location: S-473 for both Section 201 (CRN 2965) and Section 202 (CRN 2964) S-473 (2:00 p.m.)

Catalog Description: A laboratory course that demonstrates the application of concepts introduced in Chemistry 211.

Purpose:

1. To introduce you to the basic laboratory skills and careful measurements.
2. To learn to properly record and handle experimental data.
3. To provide a laboratory experience that emphasizes and reinforces the principles and concepts of chemistry introduced in your CHM 211 course.
4. To acquaint you with the quantitative thinking and procedures encountered in elementary Physical chemistry, Analytical chemistry and Biochemistry with an emphasis on the interplay between theory and experiment in science.

Required Textbook:

CHM 217: Principles of Chemistry I Laboratory (2013-2014) may be obtained from the Campus Bookstore.

Other Required Items: safety goggles, bound laboratory notebook, paper towels, lock

University Policies: This course will be conducted in accord with all applicable university policies. They can be found at: <http://www.marshall.edu/academic-affairs/policies/>

Attendance: Completion of all experiments and exams is required. Attendance is required.

Making Up a Lab: Only "Excused Absences", as defined in the MU Undergraduate catalog and policies, can be made up. The proper procedure is to notify me (by e-mail, phone, or in person) as soon as possible; any documentation (such as doctor's notes) have to be submitted directly to the Dean of Student Affairs (Dr. Stephen W. Hensley; MSC 2W38) who will then notify me (for details see the MU Undergraduate catalog).

While it is possible to allow students to make up a missed laboratory by coming into another section (during the same week), it cannot be guaranteed; early notification is absolutely essential! Since the necessary chemicals and apparatus are not available except during the scheduled time for the experiment, the labs cannot be made up in the preceding or following weeks. Labs missed due to excused absence should be made up by attending another session if possible (see above).

Note that one lab grade will be dropped in computing your score. If you miss a laboratory it will become your drop grade. If you miss more than one lab you will be give a zero on that lab if your absence was not excused. If you have excused absences for more than one lab and do not make them up during the week of the lab, you will be given an incomplete in the course so you can makeup the missed labs during a subsequent semester.

Grading: The grade in this lab will be based on a wide variety of evaluation tools including exams, quizzes, laboratory results and notebook, and Pre- and Post Lab questions. The weight for each component is as follows:

Quizzes (drop the lowest score)	20%
Midterm exam (approximately 1 hour in length)	15%
Final exam (approximately 1 hour in length)	15%
Pre- and Post- Lab Questions	20%
Experimental results and Laboratory Notebook (drop lowest score)	30%

The letter grades will then be assigned on a curve based on the average computed using the above weights.

Laboratory Notebook: For each experiment you will be required to keep records of what you did in a laboratory notebook. Most scientific writing uses the past tense and passive voice, and avoids first person statements. This should be particularly true in the experimental section of your notebook records.

Safety: An online safety training module will be available for this course on MUOnline. You are required to complete this training and pass the test at the end of the training program. Sign and bring to class the form that states you will abide by the safety rule.

Important safety rules are listed on page 5 of your lab manual. Read and obey them.

Schedule of Experiments

<u>Date</u>	<u>Exp. No.</u>	<u>Experiment</u>
1/13	1 Part Ia and b	Lab check in. Density of water.
1/27	1 Part 1b	Density of sugar water & sodas
2/3	2	Separating components of a mixture
2/10	3	Percent oxygen in air
2/17	4	Reactions
2/24	5	Determination of Avogadro's number
3/3	6	Heat of Reaction and Heat of Solution
3/10	7	Synthesis of an Alum <u>& Mid Term Exam</u>
3/24	8	Determination of an Empirical Formula
3/31	9	Titration of Vinegar
4/7	10	Combustion
4/14	11	Molecular Architecture
4/21	12	Energy in a Peanut
4/28	Lab Check Out	<u>Final Exam</u>