

Introductory Physical Chemistry / CHM 307

College/Department	Chemistry
Semester	Spring 2014
Instructor	Dr. Brian Scott Day
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Office	Science Building, 479
Location	Lecture: Science Building 465 Laboratory: Science Building 470
Days/Times	9:30 – 10:45 AM, TR 1:00 – 2:45 PM, W Section 201 3:00 – 4:45 PM, W Section 202
Office Hours	1:00 – 3:00 PM TR

Information for drop or withdraw is available on the Academic Calendar
<http://www.marshall.edu/calendar/academic/>

COURSE MATERIALS

1. *Physical Chemistry for the Biosciences* by Raymond Chang
2. Access to MU Online and a Marshall email account
3. Bound notebook (not spiral) for lab
4. Indirectly vented chemical safety goggles for lab
5. Calculator
6. ACS academic lab safety guide
http://portal.acs.org/portal/PublicWebSite/about/governance/committees/chemicalsafety/publications/WP_CP_012294

COURSE DETAILS

Description:

A brief survey of physical chemistry including the topics of thermodynamics, kinetics, molecular structure, and spectroscopy. Intended for students needing a broadly based science background. **Credit Hours** 4.00 **Prerequisites:** grade of C or higher CHM 212, C or higher in MTH 140 or 229, and 8 hours of physics, or consent of instructor

Objectives:

1. To apply concepts of thermodynamics, kinetics, molecular structure, and spectroscopy to problems of chemical significance.
2. To learn how to apply general chemical knowledge to solve basic chemistry problems and to implement these logical strategies to solve complex, multistep problems.
3. To understand and interpret scientific literature.
4. To refine oral communication skills.
5. Students will enhance their writing skills and strategies.

COURSE OUTCOMES

Student Learning Outcomes	How students will practice each outcome in this course	How student achievement of each outcome will be assessed in this course
Students will use the laws of thermodynamics to understand energy transfer among matter and to derive relationships relating energy transfer and physical observables.	<ul style="list-style-type: none"> lectures and readings on thermodynamics homework 	<ul style="list-style-type: none"> tests and quizzes lab reports
Students will understand how statistics and the energy of a system governs the rate and extent of chemical reactions.	<ul style="list-style-type: none"> lectures and readings on statistical mechanics, equilibrium, and reaction kinetics homework 	<ul style="list-style-type: none"> tests and quizzes
Students will learn the concepts of quantum mechanics and use them to explain molecular structure and spectroscopy.	<ul style="list-style-type: none"> lectures on quantum mechanics homework 	<ul style="list-style-type: none"> tests and quizzes lab reports
Students will apply algebra and calculus to solving problems of chemical and thermodynamic significance.	<ul style="list-style-type: none"> weekly recitations homework 	<ul style="list-style-type: none"> tests and quizzes
Students will understand, interpret and discuss scientific literature.	<ul style="list-style-type: none"> journal clubs 	<ul style="list-style-type: none"> students will write critical evaluations of journal articles
Students will enhance writing skills and strategies.	<ul style="list-style-type: none"> low-stakes short essays in homework sets, quizzes, and exams peer-editing of lab reports and journal article critiques editing conferences with Dr. Day 	<ul style="list-style-type: none"> lab reports lab notebook tests and quizzes journal critiques
Students will refine oral communication skills.	<ul style="list-style-type: none"> journal club discussions group work in lab 	<ul style="list-style-type: none"> group presentations

GRADING

	Percent of Grade
Homework/Quizzes	10
Lab Notebook	5
Lab Reports	20
Journal Article Critiques	10
In-Class Exams (3)	30
Final Exam	20
Final Presentation	5

A 90 -100 %, B 80 -89 %, C 70 - 79%, D 60 - 69 %, F 0 - 59 %

COURSE SCHEDULE AND ACTIVITIES

(List all projects, exams, and other course requirements. Be specific regarding descriptions of each project, and outline any special requirements. For example, if there will be a grading penalty for late papers, this should be explained.)

Dates	Chapter(s)	Lab
1/14-16	1, 2	No Lab
1/21-23	2, 3	Introduction to Laboratory
1/28-30	3	Gas laws
2/4-6	3,4	Bomb calorimetry
2/11-13	4	Bomb calorimetry
2/18-20	Test 1 (2/18); chapter 5	Computational chemistry
2/25-27	6	Journal club
3/4-6	6, 9	Clock reactions
3/11-13	9	Clock reactions
3/18-20	<i>Spring Break</i>	
3/25-27	chapter 10, Test 2 (3/27)	Journal club
4/1-3	11	Spectroscopy of conjugated dyes
4/8-10	11,12	
4/15-17	12, 14	Computational chemistry
4/22-24	14, Test 3 (4/17)	Spectroscopy of HCl/DCI
4/29-5/1	group projects	Journal club and review
5/6	8:00 AM FINAL EXAM	

Writing-Across-the-Curriculum Designation

This course has been designated "WI," or writing-intensive. Throughout the course, students will develop the ability to comprehend the concepts of physical chemistry and interpret them for others through written communication. The following course activities are examples of how students will develop their writing skills this semester.

Homework sets- each will include at least one qualitative, short-response question

Exams- approximately 20% of exam content will require written responses (without calculations)

Lab notebook- standards of industry and academia will be reinforced

Journal club- written responses to analytical critique questions

Lab reports- written in the style of research manuscripts

Group projects- powerpoint slides will be written to effectively complement the presentations

COURSE POLICIES

Attendance Policy

Attendance for this class is not mandatory, but highly recommended. Absences from quizzes, exams and laboratories can only be made-up if the absence falls within one of the categories outlined in the undergraduate catalog as an excused absence. To make-up a quiz, exam or lab, you will need to follow the process for securing an excused absence. All excused absences must be obtained as soon as possible.

Laboratory Policies

1. Students must complete the lab safety training on MU Online prior to entering the laboratory.
2. Goggles are required at all times during lab.
3. Open-toed shoes, shorts, bare midriffs, etc. are not allowed
4. Blatant disregard for standard safety practices will result in dismissal from lab

Other Course Policies

1. Homework and lab reports will not be accepted after their due dates.
2. Cell phones cannot be used, or out, during exams.
3. Sharing calculators during quizzes/exams is prohibited.
4. During quizzes/exams, all materials necessary will be provided to you except a pencil and calculator. You may NOT use your own paper, etc.
5. Please turn off cell phones during class, failure to do so may result in dismissal from lecture.
6. Class announcements may be made via email to your university email address and it is your responsibility to check that account on a regular basis.

UNIVERSITY POLICIES

Academic Dishonesty

All students should be familiar with the university's policy concerning academic dishonesty. This policy can be found on pp. 66 - 68 of the undergraduate catalog http://www.marshall.edu/catalog/undergraduate/ug_10-11_published.pdf, or on pp. 61 – 63 in the 2009 online graduate catalog http://www.marshall.edu/catalog/graduate/S2009/gr_sp09_published.pdf. (Faculty are encouraged to add any additional information specific to their expectations and/or rules regarding academic dishonesty in their class).

Policy for Students with Disabilities

Marshall University is committed to equal opportunity in education for all students, including those with physical, learning and psychological disabilities. University policy states that it is the responsibility of students with disabilities to contact the Office of Disabled Student Services (DSS) in Prichard Hall 117, phone 304 696-2271 to provide documentation of their disability. Following this, the DSS Coordinator will send a letter to each of the student's instructors outlining the academic accommodation he/she will need to ensure equality in classroom experiences, outside assignment, testing and grading. The instructor and student will meet to discuss how the accommodation(s) requested will be provided. For more information, please visit <http://www.marshall.edu/disabled> or contact Disabled Student Services Office at Prichard Hall 11, phone 304-696-2271.

University Computing Services' Acceptable Use Policy:

All students are responsible for knowing this policy, which can be found on the web at <http://www.marshall.edu/ucs/CS/accptuse.asp>.

Affirmative Action Policy:

This course will follow Marshall University's policy on Affirmative Action, which can be found on p. 63 of the undergraduate catalog http://www.marshall.edu/catalog/undergraduate/ug_10-11_published.pdf, or on pp. 16-17 of the 2008 graduate catalog http://www.marshall.edu/catalog/graduate/S2009/gr_sp09_published.pdf. Specifically, all students will be afforded equal opportunity without regard to race, color, sex, religion, age, disability, national origin, or sexual orientation.