Course Title/Number	Principles of Chemistry II / CHM 212, Section 101	
Semester/Year	Fall 2017	
Days/Time	MWF 1:00 – 1:50	
Location	465 Science Hall	
Instructor	Dr. Scott Day	
Office	479 Science Hall	
Phone	304-696-7054	
E-Mail	day17@marshall.edu	
Office/Hours	Mondays and Wednesdays 10:00 – 12:00 (W 11:00 – 12:00 will be in S-460)	
	Thursdays 10:00 – 11:00	
	Drop-in visits are welcome	
University Policies	By enrolling in this course, you agree to the University Policies listed below. Please read the full text of each policy by going to www.marshall.edu/academic-affairs and clicking on "Marshall University Policies." Or, you can access the policies directly by going to http://www.marshall.edu/academic-affairs/?page_id=802	
	Academic Dishonesty/ Excused Absence Policy for Undergraduates/ Computing Services Acceptable Use/ Inclement Weather/ Dead Week/ Students with Disabilities/ Academic Forgiveness/ Academic Probation and Suspension/ Academic Rights and Responsibilities of Students/ Affirmative Action/ Sexual Harassment	

Course Description:

A continuation of CHM 211 with emphasis on the inorganic chemistry of the representative elements and transition metals. 3.00 credits. Prerequisite: grade of C or better in CHM 211

Course Outcomes:

Course Student Learning Outcomes	How students will practice each outcome in this Course	How student achievement of each outcome will be assessed in this Course
Students will identify and explain	-lectures	-exams
trends in physical and chemical	-textbook readings	-ALEKS exercises
properties.	-ALEKS exercises	
Students will understand how the	-lectures	-exams
energy of a system governs the	-textbook readings	-ALEKS exercises
rate and extent of chemical	-ALEKS exercises	
reactions.		
Students will understand how the	-lectures	-exams
relative amounts of chemical	-textbook readings	-ALEKS exercises
species govern the rate and	-ALEKS exercises	
extent of chemical reactions.		
Students will apply mathematical	-lectures	-exams
techniques to formulate and	-textbook readings	-ALEKS exercises
solve problems in chemistry.	-ALEKS exercises	

Required Texts, Additional Reading, and Other Materials:

- 1. Principles of General Chemistry, Third Edition by Martin S. Silverberg; McGraw-Hill, 2013
- 2. ALEKS access
- 3. Access to MU Online and a Marshall email account
- 4. Non-programmable calculator

Course Policies

Grading Policy

The grade for this class will be determined from homework, four in-class exams and a cumulative, final exam. The homework portion of the grade will be determined from ALEKS exercises. One hundred points of the homework will come from completion of the periodic objective assignments and one hundred points from topic mastery (% of topics mastered from pie chart) at the end of the semester. The material for the exams will come from lectures, ALEKS problems and the reading assignments. In-class exams may cover material from previous exams.

ALEKS exercises	200 points
In-class exams	600 points
Final exam	200 points
	1000 total

Grading Scale: A 900-1000 B 800-899 C 700-799 D 600-699 F < 600

Attendance Policy

Attendance for this class is not mandatory. By that, no portion of your grade will be determined by attendance. Absences from exams can only be made-up if the absence falls within one of the categories outlined in the undergraduate catalog for excused absences. To make-up an exam, you will need to follow the process for securing an excused absence. Excused absences must be obtained as soon as possible.

Other Policies

- 1. Cell phones cannot be used, or out, during exams.
- 2. Sharing calculators during exams is prohibited.
- 3. During exams, all materials necessary will be provided to you except a pencil and calculator. You may NOT use your own paper, etc.
- 4. Please turn off cell phones during class, failure to do so may result in dismissal from lecture.
- 5. Class announcements may be made via email to your university email address and it is your responsibility to check that account on a regular basis.

Course Schedule

Week	Chapter	Notes
August 21	Syllabus, 12	
August 28	13	
September 4	16	No class on Labor Day
September 11	16	
September 18	17	Exam I (chaps. 12, 13 & 16)* F
September 25	17	
October 2	18	
October 9	18	
October 16	19	Exam II (chaps. 17 & 18)* W
October 23	19	
October 30	20	
November 6	21	Exam III (chaps. 19 & 20)* W
November 13	21	
November 20		Thanksgiving Break
November 27	22	Exam IV (chaps. 21 & 22)* F
December 4	23	
December 9		Final Exam at 10:00

^{*}Exam dates are approximate and subject to change